Polyoxometalates as a case study of super-chaotropes

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lonic species, such as polyoxometalates (POMs) or (metal-) boron clusters, are at the frontier between ions and (charged-)colloids due to their nm size. We show here that the large size and low charge density of POMs, compared to classical ions, are responsible for a peculiar behavior called "super-chaotropy". This property refers to the strong propensity of nano-ions to adsorb at neutral polar interfaces, via non-specific interactions. It has strong effects on phase transitions in soft matter and can, for example, give an explanation on the origin of the famous "POM-etherate" phase formation. A simple way for evaluating and classifying nano-ions, such as POMs, according to their super-chaotropy is proposed here. The super-chaotropic behavior of nano-ions opens opportunities in separation science, a catalysis, and for the design of nanostructured hybrid materials.

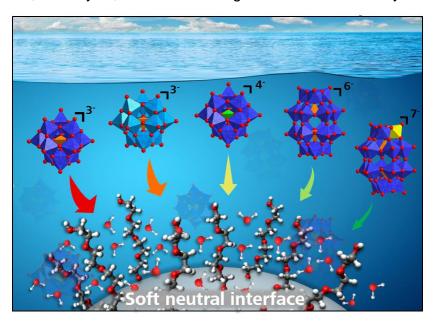


Figure 1 Nano-ions with delocalized charges, such as POMs, strongly adsorb on neutral polar interfaces. This property called "super-chaotropy" scales with the charge density and, to a lesser extent, with their polarizability.

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